Kettlethorpe HIGH SCHOOL Year 7 Knowledge Organiser Booklet

Name:



THINK PINK

If you see **PINK** in your books, make the corrections.

Capital letters

- sentence starts
 - proper nouns
 - the word 'l'

Commas

- to separate three or more items in a list
- use a pair of commas when you are inserting extra information
 - in the middle of the sentence
 - use after an adverbial

Before sunrise, Zac ate his breakfast.

Apostrophes

- to show that a letter or letters are missing: 1'm - haven't - don't
- to show something belongs to something else: The parents' meeting lasted an hour.

 Have you carefully reread your work?
 Have you checked to see if you accidentally made any mistakes?
 Are you proud of your work? Common mistakes There refers to a place or idea. Their shows belonging. They're is short for 'they are'.

use should have - not 'should of' use could have - not 'could of' use would have - not 'would of'

Spelling

- use the dictionary
- make sure to use subject specific vocabulary

APPLY THE RULES. BE CONSISTENT. CHECK FOR ACCURACY.

WWW - Descriptive comment on what went well

BI - Descriptive comment saying your work would be **even better if**

Punctuation







Do you know your roots?





root meaning 'art /skill'





MICro-

prefix meaning 'small '



root meaning '**sound** '



Specialised cells

Specialised cells are found in multicellular organisms. Each specialised cell has a particular function within the organism.

	Type of cell	Function	Special features
Animal cells	Red blood cells	To carry oxygen	 Large surface area, for oxygen to pass through Contains haemoglobin, which joins with oxygen Contains no nucleus
	Nerve cells	To carry nerve impulses to different parts of the body	 Long Connections at each end Can carry electrical signals
	Sperm cell	To reach the female cell and join with it	 Long tail for swimming Head for getting into the female cell
Plant cells	Root hair cell	To absorb water and minerals	- Large surface area
	Leaf cell	To absorb sunlight for photosynthesis	 Large surface area Lots of chloroplasts

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Key Stage 3 SCIENCE Year 7 (Term 1) | *Cells & Cell Movement*

Skeletal System

The skeletal system is made from bones and joints. The skeleton has four main functions:

- 1) to **support** the body
- 2) to **protect** some of the vital organs of the body
- 3) to help the body move
- 4) to make blood cells

Muscles

There are different types of muscle. Skeletal muscle is joined to bones. Its cells contract to make bones move and joints bend.







Joints

Joints are places where bones are able to move in different directions. Strong, cord-like tissues called **ligaments** attach to the ends of bones either side of joints. **Tendons** attach the bones in the joint to muscles which contract and relax to move them.

Antagonistic Muscle Pairs

The knee joint has two muscles that move the shin up or down. These are the hamstrings and the quadriceps. These are examples of **antagonistic muscles**.

- To pull the shin backwards, the hamstrings contract and the quadriceps relax.
- To move the shin forwards, the quadriceps contract and the hamstrings relax.





Year 7 (Term 1) Key Stage Cells ω SCIENCE ହ Cell Movement

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ettlethorpe Year 7 (Term 1) Key Stage Particle Model & ω S **SCIENCE** Separation techniques

Changes of State

Changes of state take place when particles gain or lose energy.



Distillation

Distillation separates substances with different boiling points. This is good for separating a mixture of liquids, e.g. ink and water.



<u>Solutions</u>

A **solvent** is the liquid that can dissolve a solid (e.g. water).

A solute is the solid that can dissolve in a liquid (e.g. salt or sugar). During dissolving, the solvent particles

surround the solute particles and move them away, so they spread out in the solvent. This makes a **solution**.



Filtration

Filtration is the process which removes small insoluble particles from liquid. Filter paper is used to separate the insoluble solid from the liquid.



Chromatography

Chromatography is used to separate the compounds in a mixture according to how soluble they are.





Diffusion

Diffusion is how smells spread out through the air and how concentrated liquids spread out when placed in water.

Diffusion happens on its own when the particles spread out from **an area of high concentration**, where there are many of them, **to areas of low concentration** where there are fewer of them.





Cooling Curves

Changes of state can be investigated by measuring the temperature as a substance changes state.



The temperature stays the same during a state change:

- during melting and evaporating, internal energy increases as the motion of particles increases and bonds are broken
- during condensing and freezing, internal energy decreases as the motion of particles decreases and new bonds are formed

Gas Pressure

Gas pressure is caused when gas particles hit the walls of their container. The more often the particles hit the walls, and the faster they are moving when they do this, the higher the

pressure.

If a gas is **heated**, its particles move around **more quickly**. They hit the walls of their contained harder and more often, which increases the pressure.



Particles of gas collide with each other and with their container walls.

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Reflection of light

The diagram shows a ray of light reflected from a plane mirror.



When light is reflected from a plane mirror:

the angle of incidence, i = the angle of reflection, r

This is known as the law of reflection.



This diagram shows the amplitude and wavelength of a wave.

The Eye

The eye focuses light from an object onto a photo-sensitive material. However, in the eye, this material is the **retina**. The retina contains cells that are sensitive to light. They produce **electrical impulses** when they absorb light. These impulses are passed along the **optic nerve** to the **brain**, which interprets them as vision.



Hearing Sounds

We can detect sound using our ears. An ear has an **eardrum** inside, connected to three **small bones**.

The vibrations in the air make the eardrum vibrate, and these vibrations are passed through the three small bones (called ossicles) to a spiral structure called the cochlea.

Signals are passed from the cochlea to the brain through the **auditory nerve**, and our **brain** interprets these signals as sound.

Seeing Colour

The light that we see is made up of many different colours.

There are three **primary colours**: red, green and blue. If all three are mixed together we see **white light**. They can be combined in different ways to make every other colour.

The colour of an object is determined by the wavelength of light that it reflects.





Key Stage 3 SCIENCE Year 7 (Term 1) | Light & Sound

What is energy?

Energy is a quantity that is conserved - it cannot be created or destroyed. Energy can be stored and transferred. There are several **stores** of energy.

Kinetic energy store

The amount of energy in the kinetic energy store depends on the **speed** of the object.

Gravitational potential energy store

The amount of energy in the gravitational potential energy store depends on the **height** of the object.

Thermal energy

The amount of energy in the thermal energy store depends on the **temperature** of the object.

Chemical energy

Batteries, foods and fuels store energy in their chemical energy stores. Transfer of energy from the chemical energy store occurs due to chemical reactions.

Elastic potential energy

The amount of energy in the elastic energy store depends on the amount of extension or compression.

Law of Conservation of Energy

The law of conservation of energy state: "energy cannot be created nor destroyed, only transferred from one store to another".

Generating electricity

Much of the energy that is transferred in our homes is supplied by electricity. There are a wide range of energy resources used to generate electricity.

Energy resources are systems that can store large amounts of energy. Energy resources can be divided into two categories: 1. **Renewable** resources - energy resources that can be replenished. They do not run out although we are using them. 2. **Non-renewable** resources energy resources that cannot be replenished. These resources will eventually run out because we are using them. ket

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Reactions of Metals with Acids Acids react with most metals. When an acid reacts with a metal, the products are a salt and hydrogen.

This is the general word equation for the reaction:

metal + acid \rightarrow salt + hydrogen



Displacement

Displacement reactions involve a metal and the compound of a different metal. A more reactive metal will displace or push out a less reactive metal from its compound in a displacement reaction. The less reactive metal is left uncombined after the reaction. It is no longer chemically bonded to any other elements. It is now a pure element.

Most reactive		Reaction with dilute acids
Potassium		Violent reaction
Sodium		Violent reaction
Calcium	00	Rapid bubbling
Magnesium	00	
Aluminium	<u> </u>	Rapid bubbling but slow at first
Zinc	о о По	
Iron		Slow bubbling
Tin		
Lead 6		Very slow bubbling
Silver		No reaction
Gold		
Platinum		
Least reactive		

Reactions of Metals Some metals are very reactive. This means they easily take part in chemical reactions to make new substances. Other metals are very unreactive, and do not easily take part in chemical reactions. If we put the metals in order of their reactivity, from the most reactive down to the least reactive, we get a list called the Reactivity Series (left).



Neutralization

A chemical reaction that takes place between an acid and an alkali is called a neutralization reaction. It produces a salt and water. See diagram below.



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Acids

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Using an indicator An indicator is a substance which will change colour depending on the pH of the solution it is mixed with. Examples of indicators are:

- Litmus
- Universal indicator
- Phenolphthalein
- Red cabbage



Puberty

Puberty is the stage in life when a child's body develops into an adult's body. Changes occur at puberty because of hormones - testosterone in boys and oestrogen in girls.





Key Stage 3 SCIENCE Year 7 (Term 2) | Human *Reproduction*

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Sedimentary Rocks

A river carries, or **transports**, pieces of broken rock as it flows along. When the river reaches a lake or the sea, the rocks are **deposited**. See diagram for sedimentary rock formation.

PRESSURE

NCREASING

Examples of sedimentary rocks are:

- Chalk
- Sandstone
- Shale
- Mudstone





Metamorphic Rocks

Metamorphic rocks are formed from other rocks that are changed because of heat or pressure. They are not made from molten rock rocks that do melt form igneous rocks instead.

As a result, the rocks are heated and put under great pressure. They do not melt, but the minerals they contain are changed chemically, forming metamorphic rocks.

Examples of metamorphic rocks

Metamorphic Rock Formation







Year Key (Term Sta 00 P ω ω Earth S CIENCE Structure

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Igneous rocks are formed from molten rock that has cooled and solidified. When the magma cools enough, it solidifies and

Igneous rocks contain randomly arranged interlocking crystals.

Extrusive igneous rocks form from magma that erupted onto the surface as lava, where it cooled guickly. On the other

hand, intrusive igneous rocks form from magma that cooled slowly, deep underground.

Solar System

The solar system consists of the Sun, with planets and smaller objects such as asteroids and comets in orbit around it. Starting with Mercury, which is the closest to the Sun, the planets are:

- Mercury
- Venus
- Earth
- Mars
- Jupiter
- Saturn
- Uranus
- Neptune

This sentence is a way to remember the correct order:

My Very Easy Method Just Speeds Up Naming.

Day and night

The Sun lights up one half of the Earth, and the other half is in shadow. As the Earth spins, we move from shadow to light and back to shadow and so on.





Gravity

Gravity is a force that attracts objects towards each other. The more mass an object has, the greater its force of gravity:

- gravity forces between the Earth and the Moon keep the Moon in orbit around the Earth
- gravity forces between the Sun and the Earth keep the Earth in orbit around the Sun

Mass

The **mass** of an object is the amount of **matter** or 'stuff' it contains. The more matter an object contains, the greater its mass. Mass is measured in **kilograms**, **kg**.

Seasons

We get

different **seasons** (winter, spring, summer and autumn) because the Earth's axis is tilted. This is how it works:

- it is summer in the UK when the **Northern Hemisphere** is tilted towards the Sun
- it is winter in the UK when the northern hemisphere is tilted away from the Sun



Weight

The **weight** of an object is the gravitational force between the object and the Earth. The weight of an object depends upon its mass and the gravitational field strength.



Key Stage 3 SCIENCE Year 7 (Term 3) Space

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the duster becomes **negatively charged** and the rod becomes positively charged



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Adaptation of a camel to arid conditions



CLASSIFICATION OF ANIMALS



Classification There are millions of different types of living organisms. Scientists classify things to make it easier to study them.

Adaptations

Adaptations are features of living organisms that help them survive. These can be to do with their physical appearance - structural adaptations -Or they can be **behavioural** adaptations, which affect what the organisms do.

Adaptations of a polar bear to **Arctic conditions**

Small head White fur and ears Compact body shape

Thick layer of fat

Thick layer of fur

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Food Chains

A food chain is a list of organisms in a habitat that shows their feeding relationship, i.e what eats what. The organisms are joined by arrows which show the transfer of energy in food between them.



Food chains always start with a **producer**. This is usually a green plant or algae that completes photosynthesis to store energy from sunlight as glucose.

A primary consumer eats a producer. The rabbit is the primary consumer in the example food chain. This is in turn eaten by a **secondary consumer**, which is the fox.

After this might be a **tertiary consumer** (which eats a secondary consumer) and possibly a **quaternary consumer** (which eats a tertiary consumer).

Food Webs

Most populations of organisms that live in a habitat usually have more than one food source.

They usually consume more than one organism from the trophic level below.

This means that there are almost always more than one **food chain**, and these are interlinked into a **food web**.



Bioaccumulation

Toxic materials are poisonous. Some quickly break down into harmless substances in the environment. Others are persistent (they stay in the environment and do not break down). These substances in the food chain and damage the organisms in it, particularly in the **predators** at the end of the chain.





Pollination

Pollination is the transfer of pollen from the anthers of one flower to the stigma of another flower (of the same species).

In wind pollination, the wind carries the pollen from the anthers to the stigmas.

In insect pollination, insects carry the pollen from the anthers to stigmas (e.g. bees when they collect nectar pick up pollen and transfer it).

Burdock fruit.

Hooks

catch

animals'

coats.

Horse

fruit.

Chestnut





Fertilisation

After pollination, the pollen makes a pollen tube down the style to the ovary. When the pollen nucleus meets and fuses with the ovule, this is fertilisation.

A seed is then formed, and the ovary walls can become a fruit, depending on the plant species. A new plant will grow once the seed germinates in the correct conditions.

Seed Dispersal is Scattering Seeds

Seeds are dispersed or spread out so that they can grow without too much competition from each other. Here are some ways in which the seed can be dispersed:





4) Drop and Roll

The heavy fruit falls down from the tree. It splits when it hits the ground and the seeds roll out.

The seeds then tend to be further dispersed by animals.

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